



Hindsight is 20/20: Lessons Learned from Reactive Chemical Incidents

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- 25+ years in Process Safety Consulting
- PHA/HAZOP/LOPA facilitator, NFPA code compliance, PSM/RMP audits & PSM program development
- Specialty experience in reactive chemical hazards, combustible dust, flammable liquids, UN/DOT transportation of dangerous goods and emergency relief system design for reactive systems



Agenda

Incident Reviews

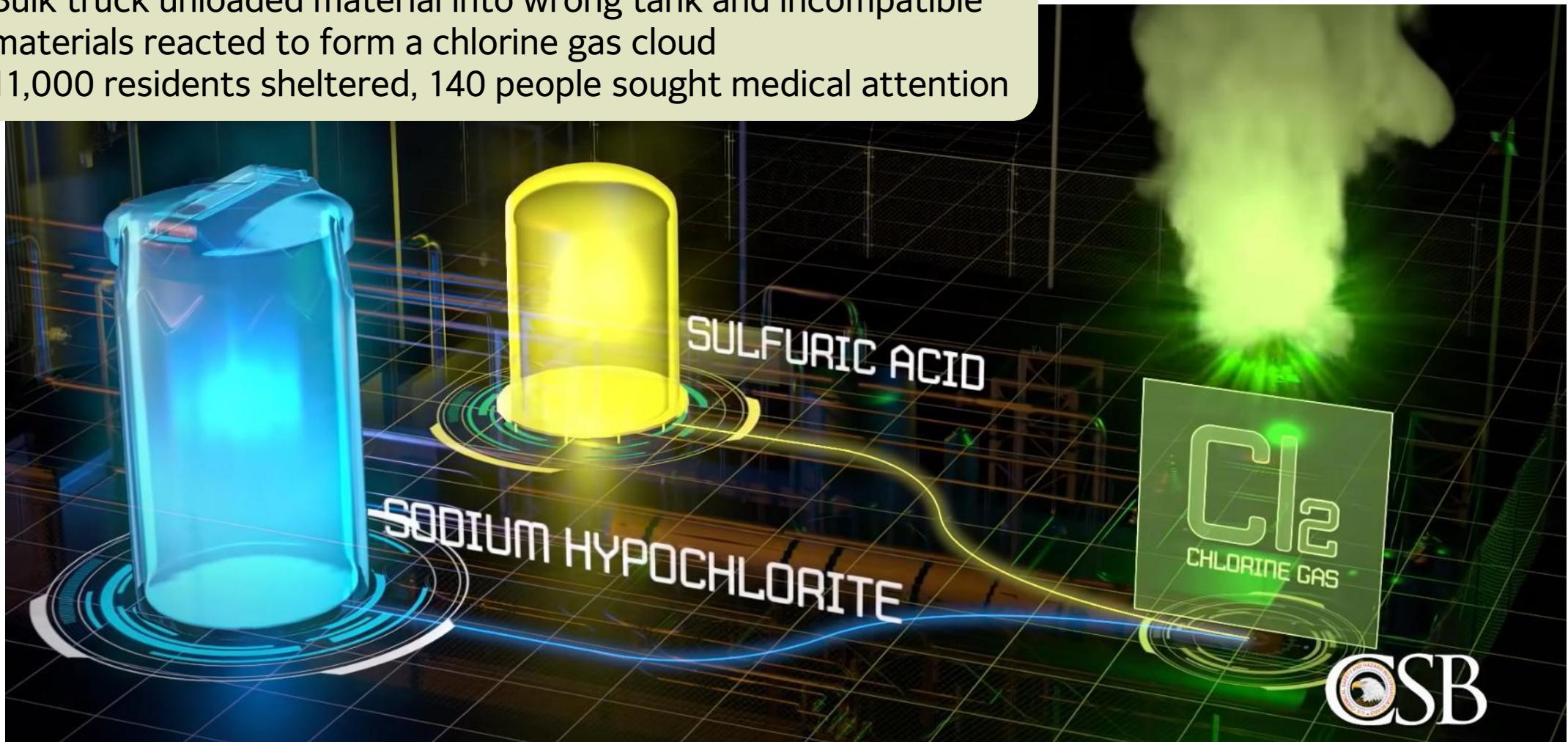
Common Causes

Reactivity Management Strategy

Key Takeaways

Storage Incident at MGPI

- Bulk truck unloaded material into wrong tank and incompatible materials reacted to form a chlorine gas cloud
- 11,000 residents sheltered, 140 people sought medical attention



Credit: US CSB Video MGPI

Storage Incident at MGPI

Human Factors

- Labeling
- Proximity

Equipment

- No automated or remote shut off valves
- Design of chemical transfer equipment

Training

- Chemical unloading procedures
- Emergency planning

Process Incident at AB Specialty Silicones

- Incorrect material loaded into tank and incompatible materials reacted to form hydrogen which was ignited, resulting in an explosion and fire
- 4 employees were fatally injured, significant damage to production area



Credit: US CSB Video AB Specialty Silicones

Process Incident at AB Specialty Silicones

Policies &
Procedures

Equipment

- Hazard analysis program was not well implemented
- Inadequate methods for storage and handling of incompatible materials
- Process safety culture weakness

- Inadequate ventilation system design
- Lack of gas detection and alarm system

Common Issues



Did not involve chemicals covered by OSHA PSM or EPA RMP



Failure to identify hazards and consequences for chemical reactivity



Procedures and training



Evaluation and design of safeguards

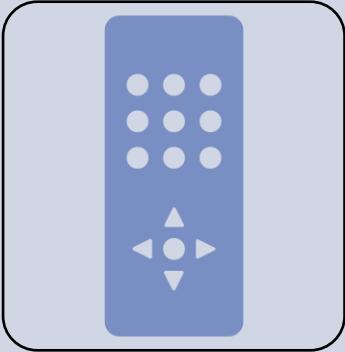


Equipment and building design concerns



Emergency preparedness

Reactivity Management Strategy



Identify

- Collect reactivity hazard information
- Identify chemical reactivity hazards

Evaluate

- Test for chemical reactivity
- Assess chemical reactivity risks

Control

- Identify process controls and safeguards
- Document chemical reactivity risks
- Communicate and train

Maintain

- Investigate incidents
- Review
- Audit
- Manage change
- Continuous improvement

Identify Reactivity – Define the Scope



Process Vessels

Intentional Chemistry

Batch Additions

Blending

Storage Tanks

Tank Farms

Loading/Unloading

Manifolds

Collect Reactivity Information

SDS review

Manufacturer technical
information

Wiley's Guide to
Chemical
Incompatibilities

Chemical reactivity
worksheet/CAMEO

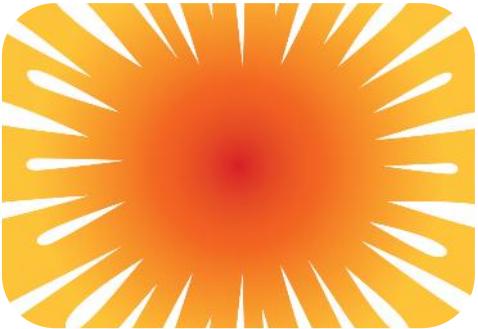
Bretherick's, Handbook
of Reactive Chemical
Hazards

Sax, Dangerous
Properties of Industrial
Materials

Single Component Hazards



Flammability



Polymerizable
(self-reactive)



Strong Oxidizing
Agent



Water-Reactive



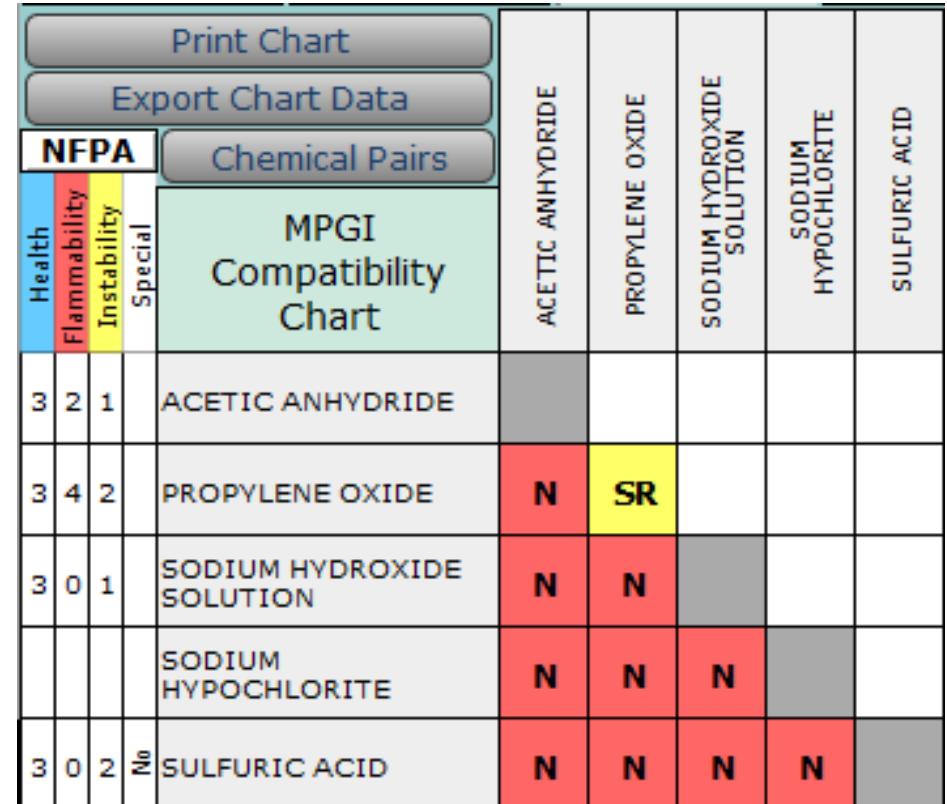
Pyrophoric



Peroxidizable
Compound

Chemical Interactivity Hazards

- Chemical Reactivity Worksheet/CAMEO
- Results can be conservative
- Create for each scope
- Limited to 1:1 interactions
- Use as first pass screening



The image shows a screenshot of the MPGI Compatibility Chart. The chart has a header with buttons for 'Print Chart', 'Export Chart Data', and 'Chemical Pairs'. The 'Chemical Pairs' button is highlighted. The chart itself has a legend for NFPA hazard ratings: Health (blue), Flammability (red), Instability (yellow), and Special (grey). The chart displays compatibility between five chemicals: ACETIC ANHYDRIDE, PROPYLENE OXIDE, SODIUM HYDROXIDE SOLUTION, SODIUM HYPOCHLORITE, and SULFURIC ACID. The data is presented in a grid where each cell contains an NFPA rating or the letter 'N' (Not Applicable). The 'N' cells are highlighted in red. The 'SODIUM HYDROXIDE SOLUTION' row has a 'Special' rating of '1' in the first column.

NFPA				Chemical Pairs	MPGI Compatibility Chart				
Health	Flammability	Instability	Special		ACETIC ANHYDRIDE	PROPYLENE OXIDE	SODIUM HYDROXIDE SOLUTION	SODIUM HYPOCHLORITE	SULFURIC ACID
3	2	1		ACETIC ANHYDRIDE					
3	4	2		PROPYLENE OXIDE	N	SR			
3	0	1		SODIUM HYDROXIDE SOLUTION	N	N			
				SODIUM HYPOCHLORITE	N	N	N		
3	0	2	No	SULFURIC ACID	N	N	N	N	

Credit: Export from AIChE Chemical Reactivity Worksheet

Develop Reactivity Scenarios

Loss of utilities

Process upset

Energy input variation

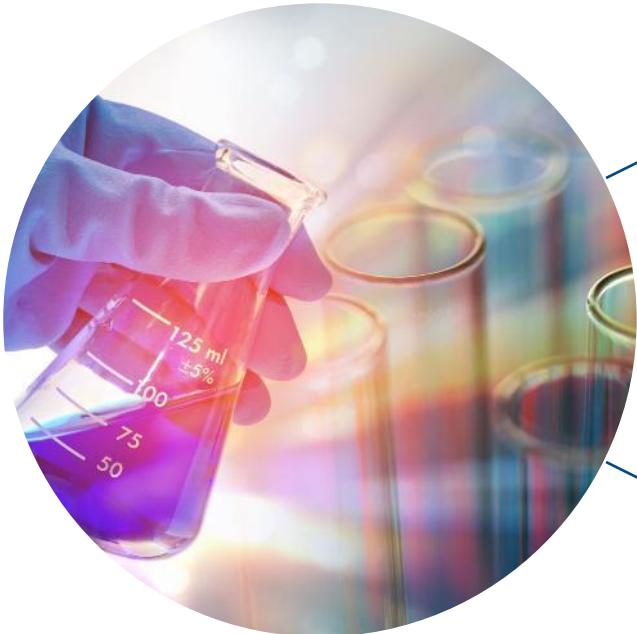
Mechanical failure

Inadvertent mixing

Human error

Other

Perform Testing to Characterize Reactivity



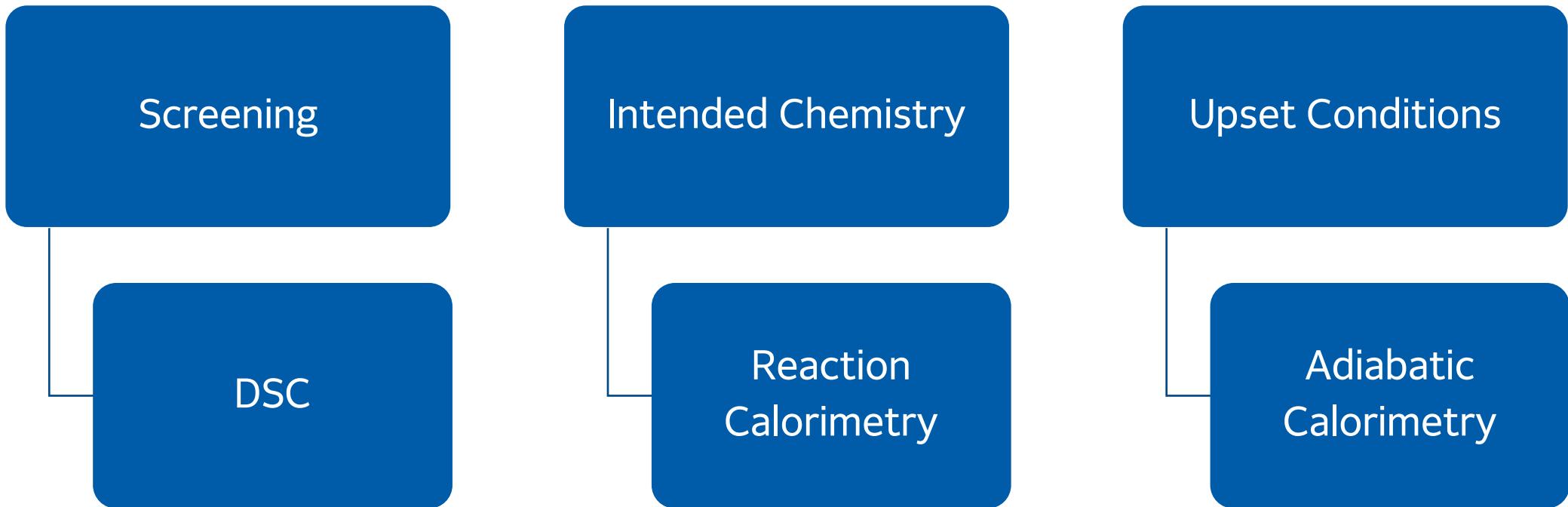
Gas production

- Flammable or toxic gas
- How much
- How fast

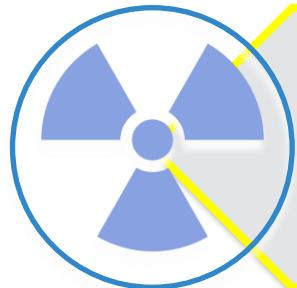
Energy release

- How much
- How fast

Testing Methodology



Evaluate Consequences



Toxic Gas Release



Flammable - Explosion



Flammable - Fire

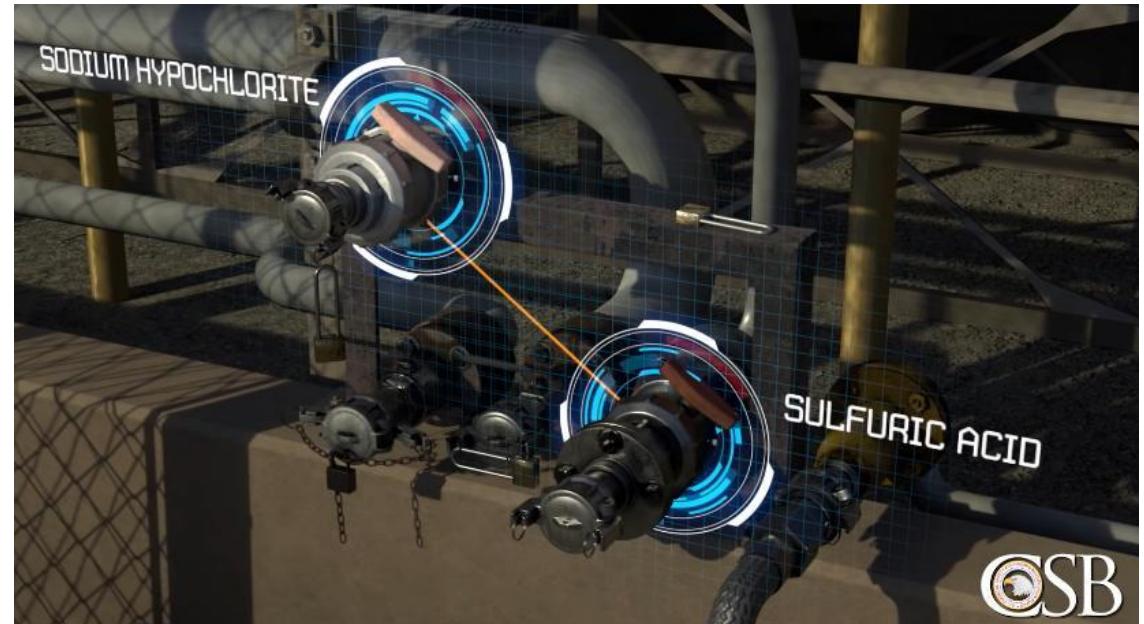
Process Hazard Analysis: Evaluate the Risk

- Upset scenario: Identify credible scenarios – be specific
- Severity: What is the worst potential consequence
- Safeguards: Identify current preventive or mitigative measures

- Evaluate: are current safeguards adequate to minimize risk to an acceptable level

Human Factors

- Field
 - Accessibility, clarity, layout
- Control room
 - Displays, alarms
- Labeling
 - Clear, consistent
- Procedures
 - Clear, consistent
- Workload
 - Regular & emergency operations



Credit: US CSB Video MGPI

Weather Extremes

- Flooding
- Tornado
- Hurricane
- Earthquake
- Cold Weather
- Hot Weather



Identify and Implement Safeguards

Interlocks

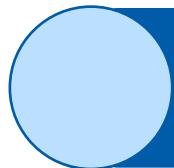
Gas Detection/Alarm

Emergency Relief System Design

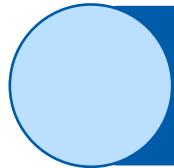
Equipment Design Considerations

Procedures

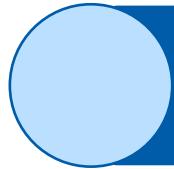
Maintain the Reactivity Program



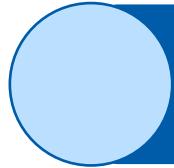
Investigate Incidents



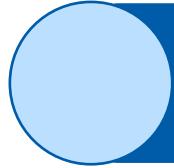
Management Review



Audit



Manage Change



Continuous Improvement

Reactivity Hazard Management Summary

- Identify reactivity hazards
- Evaluate consequences
- Control with safeguards
- Maintain the reactivity program



**HEADQUARTERS**

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